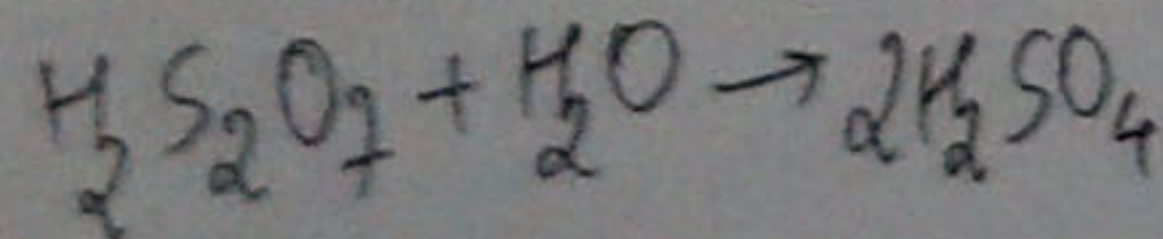
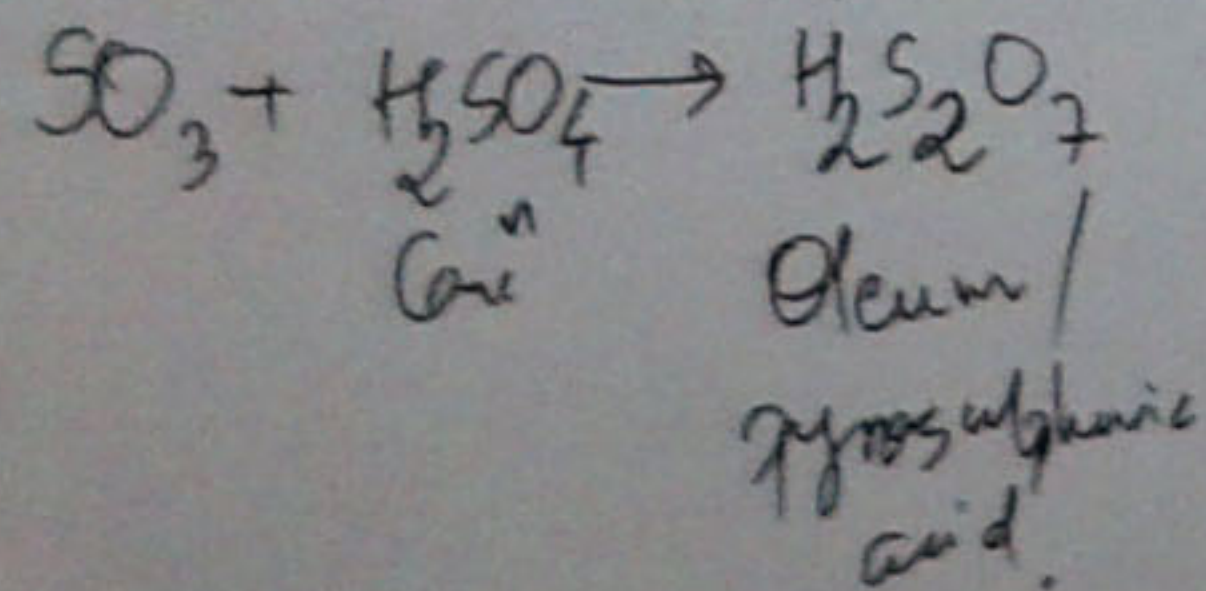
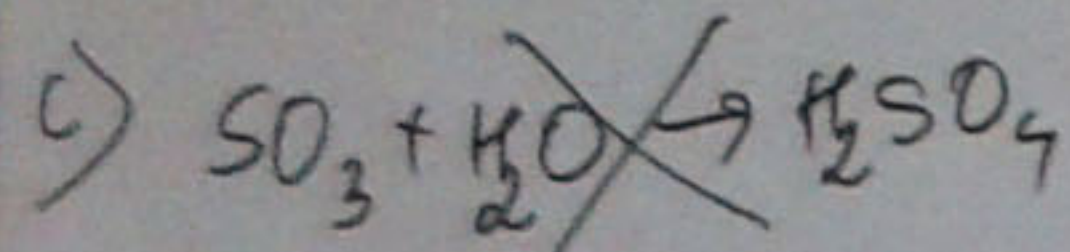
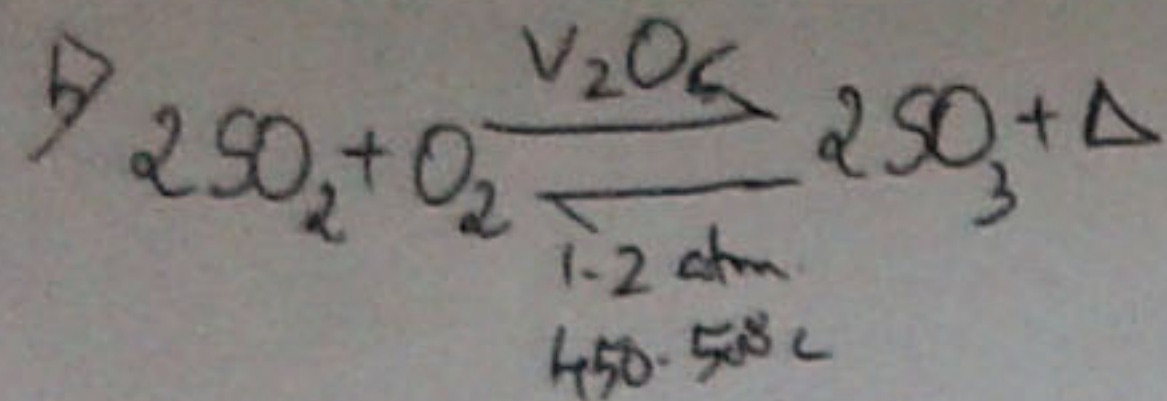
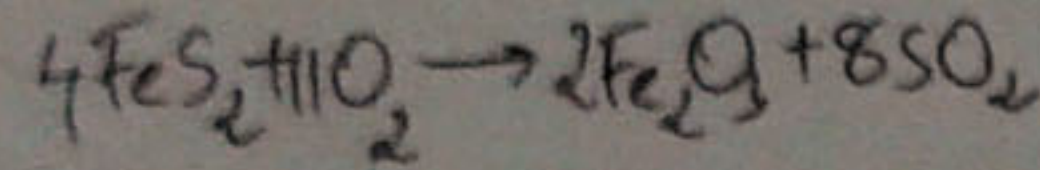
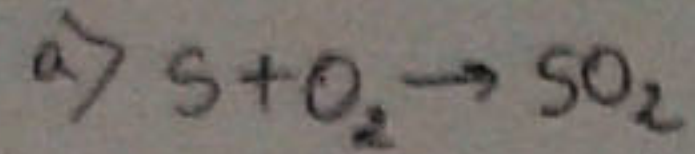
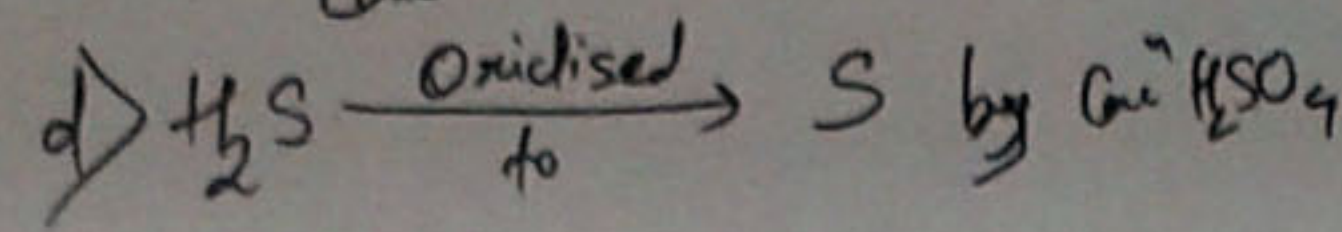
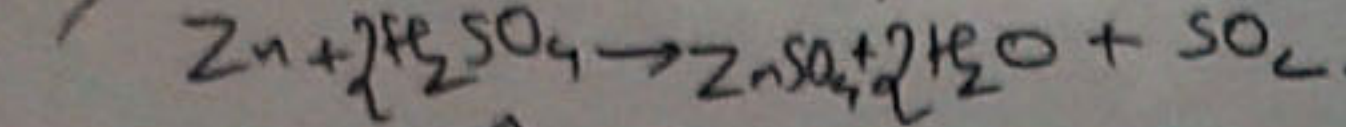
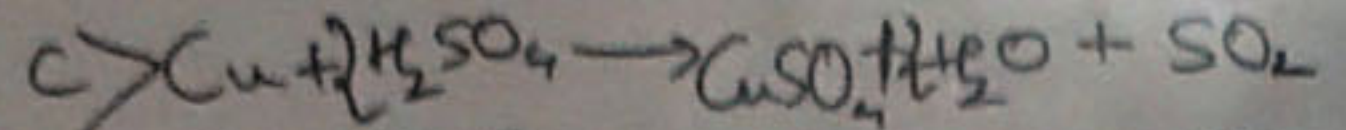
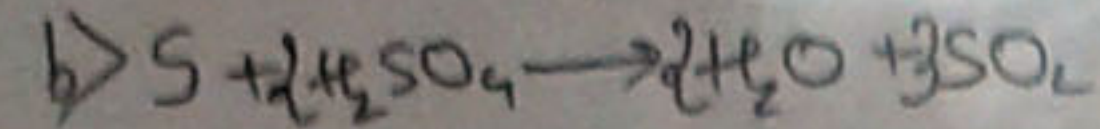


Oil of Vitriol.

Industry \rightarrow Contact process.

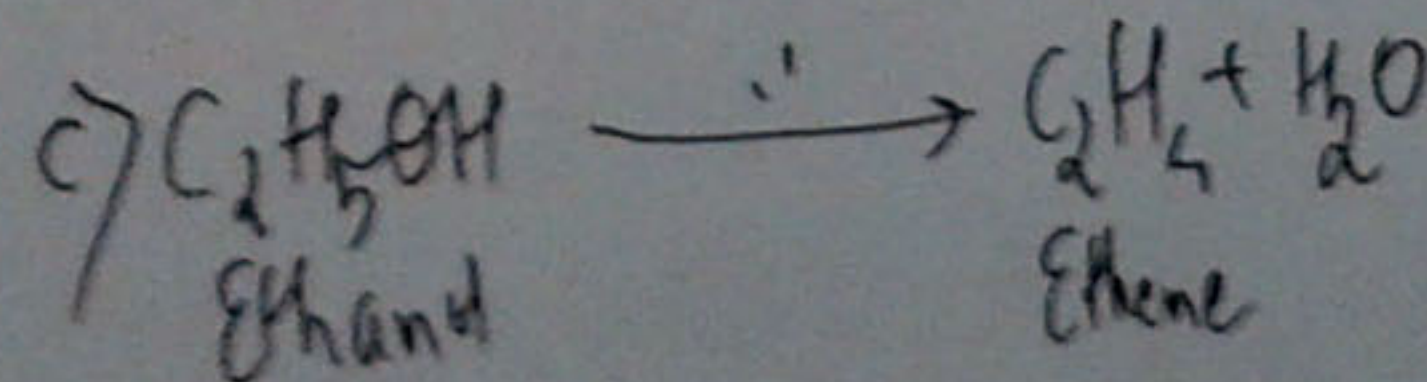
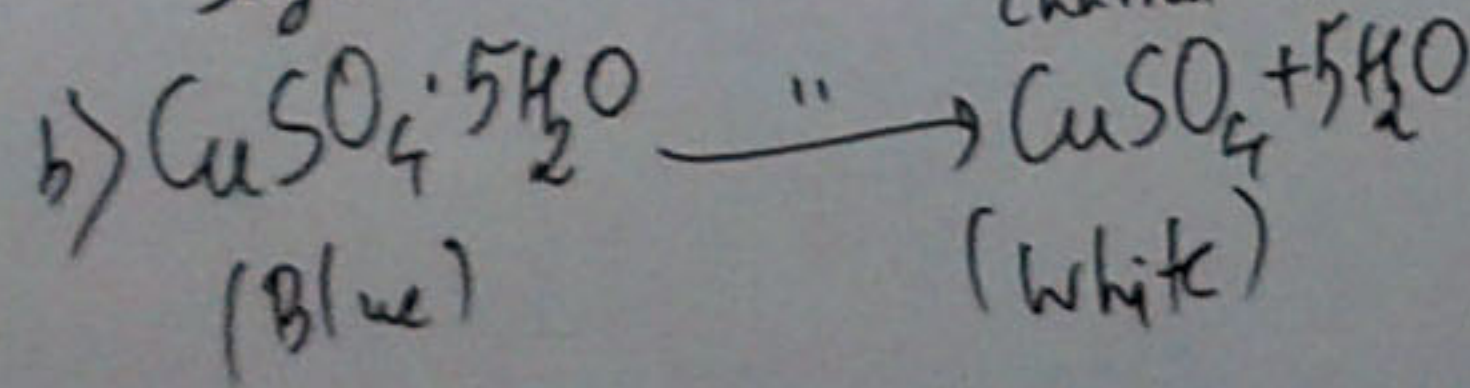
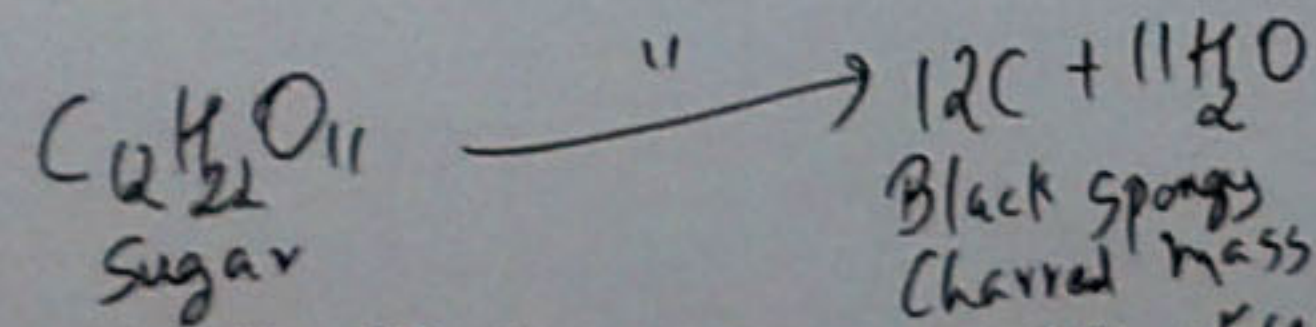
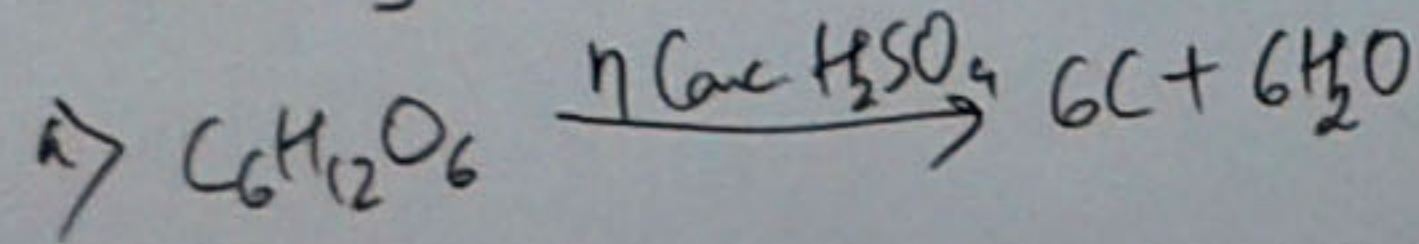


Conc H_2SO_4 is an oxidising agent:-
Justify:-

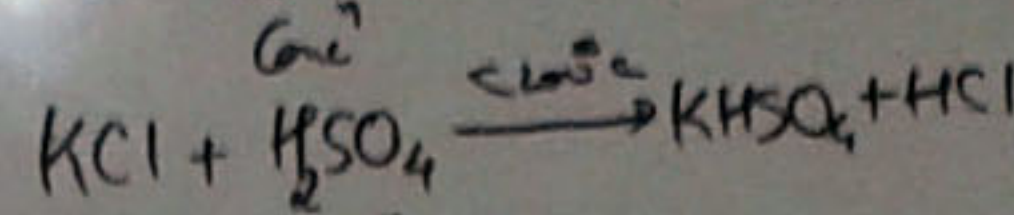
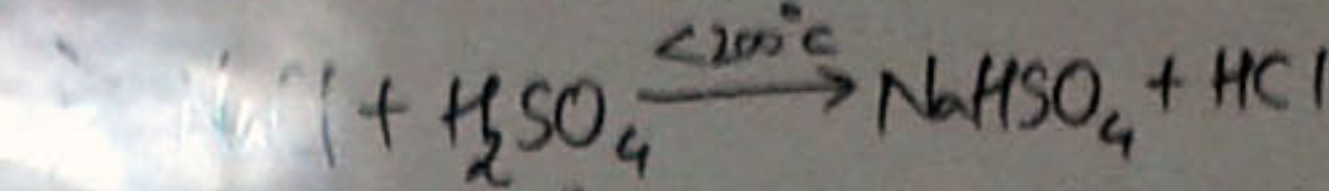


Conc H_2SO_4 is a dehydrating agent:-
becoz it has a strong affinity for water.

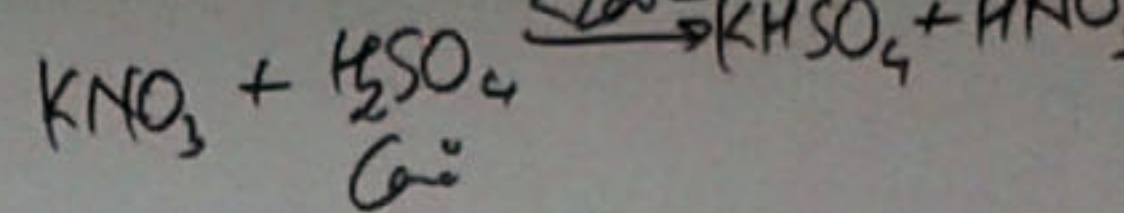
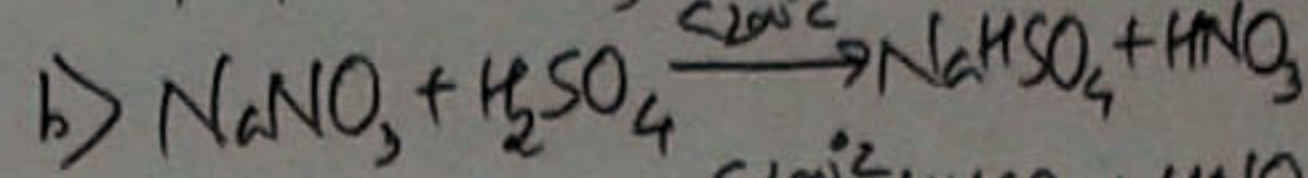
Justify:-



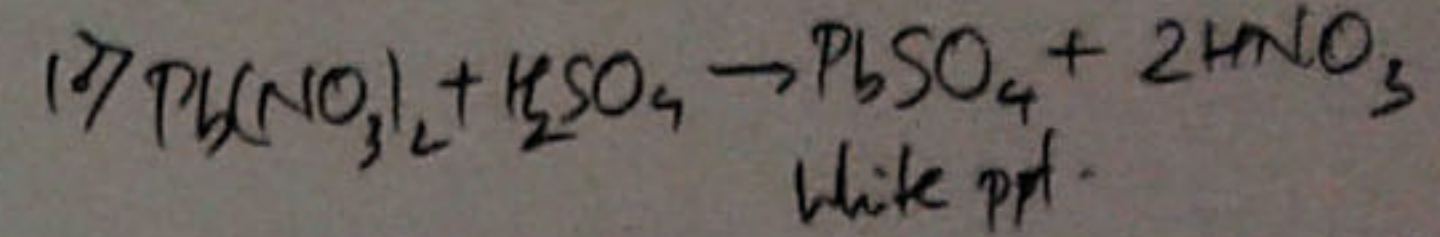
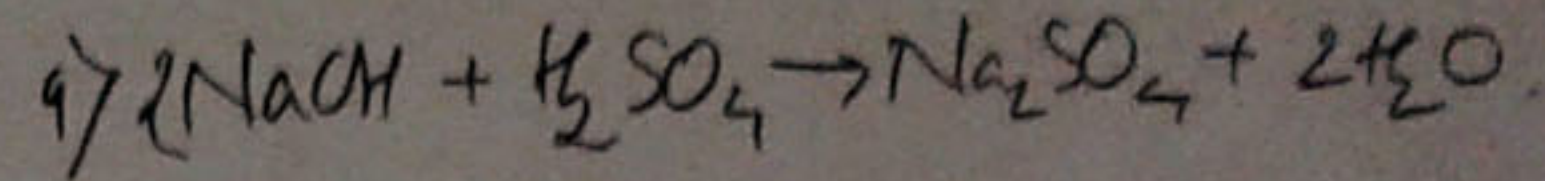
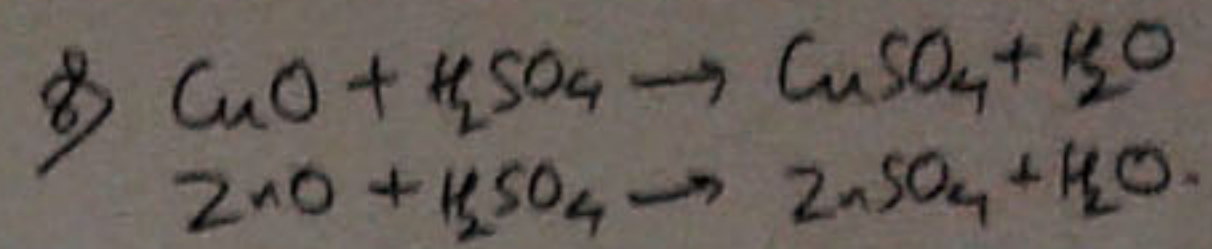
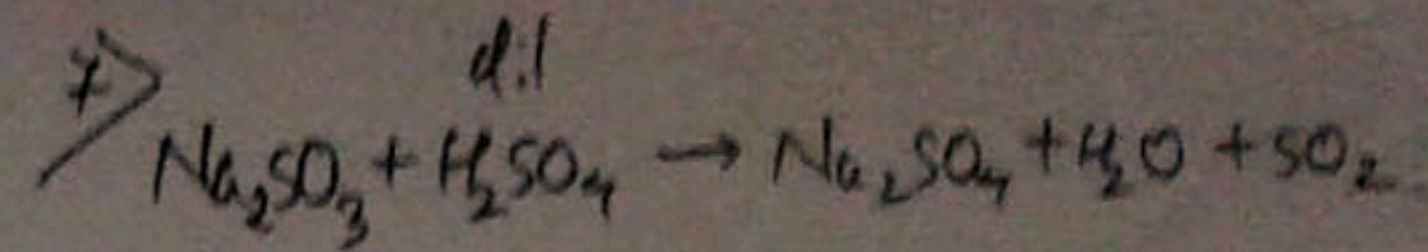
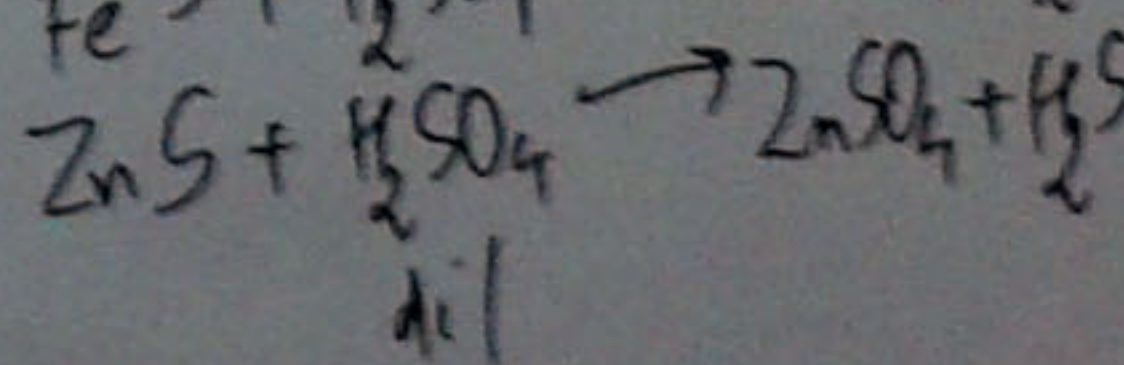
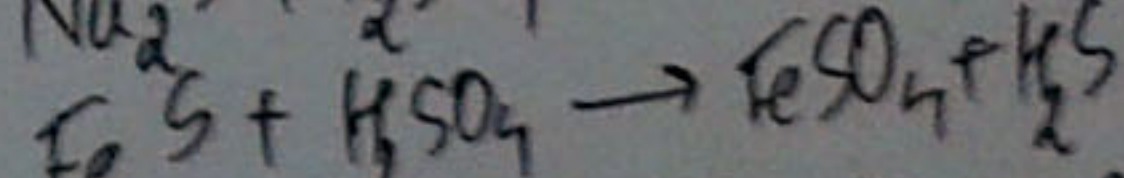
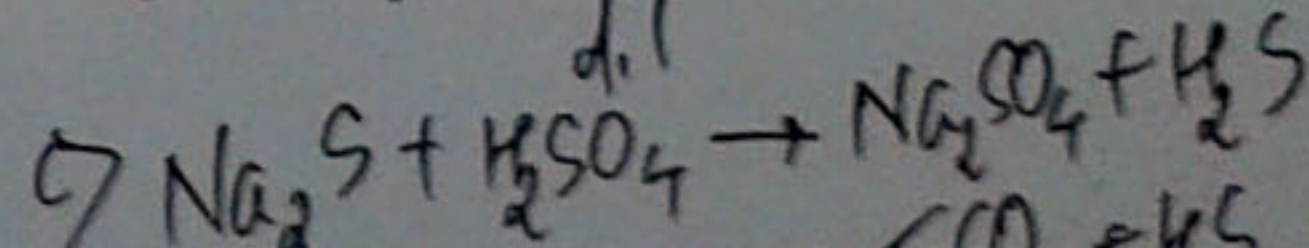
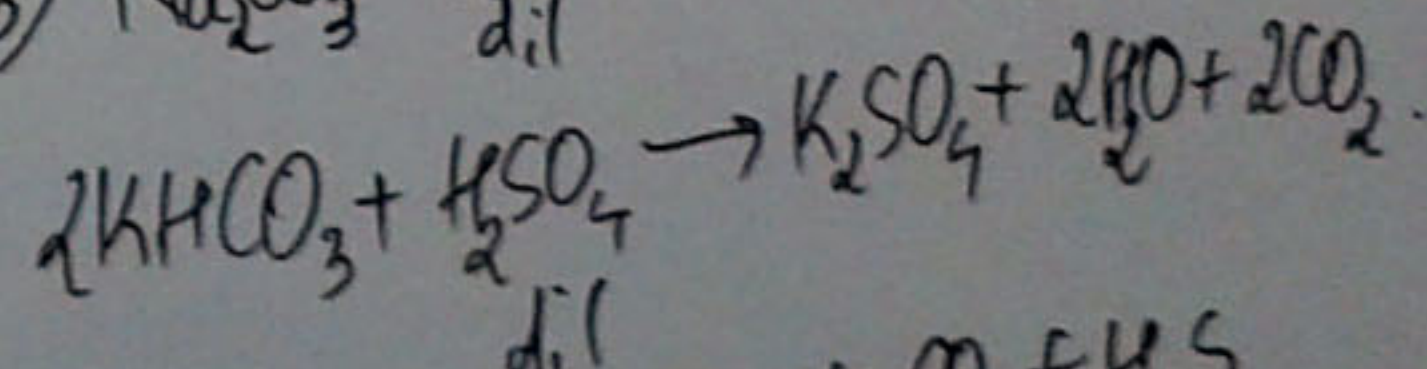
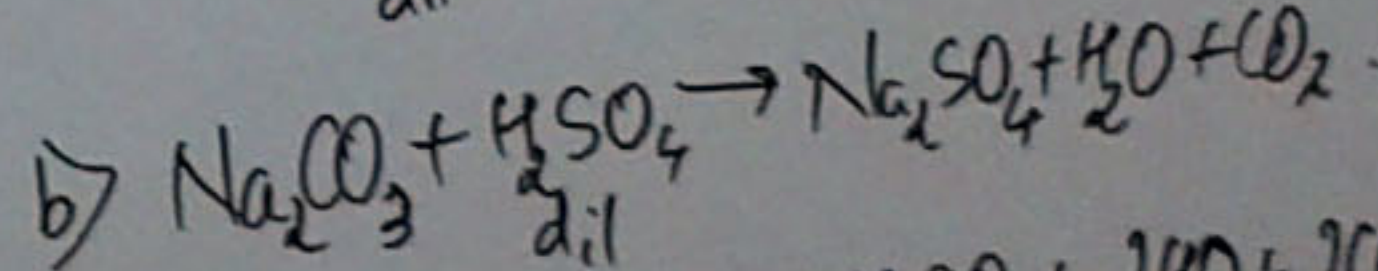
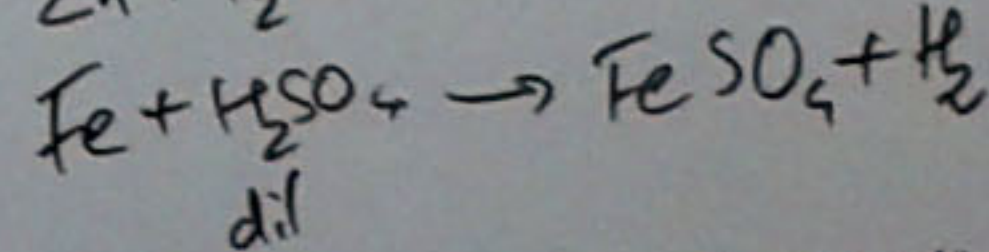
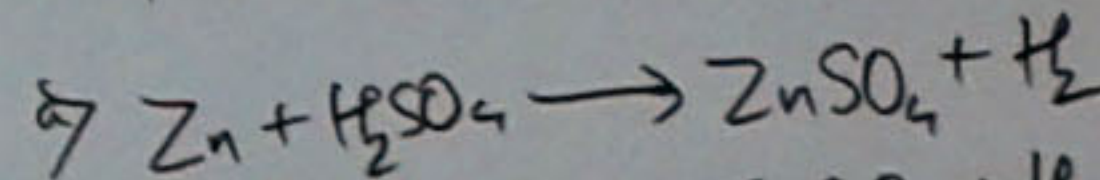
(high bp.)
Conc H_2SO_4 is a non-volatile acid.
Justify:- prepⁿ of HCl:-



prepⁿ of HNO_3 :-



Dil. H_2SO_4 shows acidic properties and it is dibasic acid



dil. HCl, HNO_3 & dil. H_2SO_4
Add $BaCl_2$

